SHORT COMMUNICATIONS

Contributions intended for publication under this heading should be expressly so marked; they should not exceed about 1000 words; they should be forwarded in the usual way to the appropriate Co-editor; they will be published as speedily as *possible.*

Acta Crvst. (1983). A39, 422

Conditions for direct structure imaging in silicon carbide polytypes. Erratum. By DAVID J. SMITH and M. A. O'KEEFE, *High Resolution Electron Microscope, University of Cambridge, Free School Lane, Cambridge CB2 3R Q, England*

(Received 11 *February* 1983)

Abstract

As a result of a printer's error, Figs. $9(a)$ and $9(b)$ in Smith & O'Keefe *[Acta Cryst.* (1983), A39, 139-148] have been transposed, so that (a) is labelled (b) and (b) is labelled (a) . All information is given in the *Abstract*.

Acta Cryst. (1983). A39, 422--424

A cusp-constrained scattering factor for bonded hydrogen atoms. By RoB J. VAN DER WAL and ROBERT F. STEWART.* *Department of Chemical Physics, University of Groningen, Nijenborgh* 16, 9747 *AG Groningen, The Netherlands*

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Abstract

A simple one-parameter form factor that satisfies a cusp condition has been developed. A one-parameter form factor, based on a Is-type density function, is more tractable in a least-squares analysis than the corresponding cusp-constrained form factor. Application of both form factors for bonded hydrogen atoms to X-ray diffraction data of sucrose revealed a more contracted density for H bonded to C compared to H bonded to O.

At a NATO-sponsored international school on electron and magnetization densities in molecules and crystals (Becker, 1980) held in Arles, France, in August 1978, a number of exercises were distributed by the participants. One exercise, proposed by Professor Hirshfeld, was to develop an atomic form factor which satisfies the electron-nuclear cusp condition. The present communication is an approximate solution to the problem for a bonded hydrogen atom.

The electron-nuclear cusp condition,

$$
\lim_{r_i \to 0} \left[(\partial \bar{\rho} / \partial \mathbf{r}_j) / \bar{\rho} \right] = -2Z_j,
$$
 (1)

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is a consequence of satisfying the many-electron *nonrelativistic* Schrödinger equation (Kato, 1957; Steiner, 1963; Pack & Brown, 1966). In (1) $\bar{\rho}$ is the one-electron density spherically averaged about the *j*th nucleus with atomic number Z_i . The r_i is a scalar radial length in atomic units (1 a.u. = 0.529177 Å). One possible form for ρ is

$$
\mu(\mathbf{r}) = A_o e^{-\gamma \mathbf{r}} \sum_{n=0}^{N} (A_n / A_o) \mathbf{r}^n / 4 \pi.
$$
 (2)

For (2) to satisfy (1) ,

$$
A_1/A_0 = \gamma - 2Z. \tag{3}
$$

For $\rho(r)$ normalized to unity,

$$
A_o = 1/[(2/\gamma)^3 (1 - 3Z/2\gamma) + \sum_{n=2}^{N} (A_n/A_0) (n+2)!/\gamma^{n+3}].
$$
\n(4)

To maintain simplicity, we set all A_n/A_0 equal to zero for $n \ge 2$. This gives a simple one-parameter function which normalizes to one and satisfies (1):

$$
\rho(\gamma, r) = (\gamma/2)^3 (1 - 3Z/2\gamma)^{-1} e^{-\gamma r} [1 + (\gamma - 2Z) r]/4\pi.
$$
 (5)

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